

Key Laboratory of Metallurgical Separation Science and Engineering

Selective co-extraction of Ni&Co from

high Ca/Mg solutions

Associate Prof./Dr. Shengxi Wu

E-mail: shengxiwu@hydrometaltech.com/csuwushengxi@126.com

Key Laboratory of Metallurgical Separation Science and Engineering, Chinese Non-Ferrous Industry

Association, Institute of Rare Metal Metallurgy research, School of Metallurgy and Environment, Central

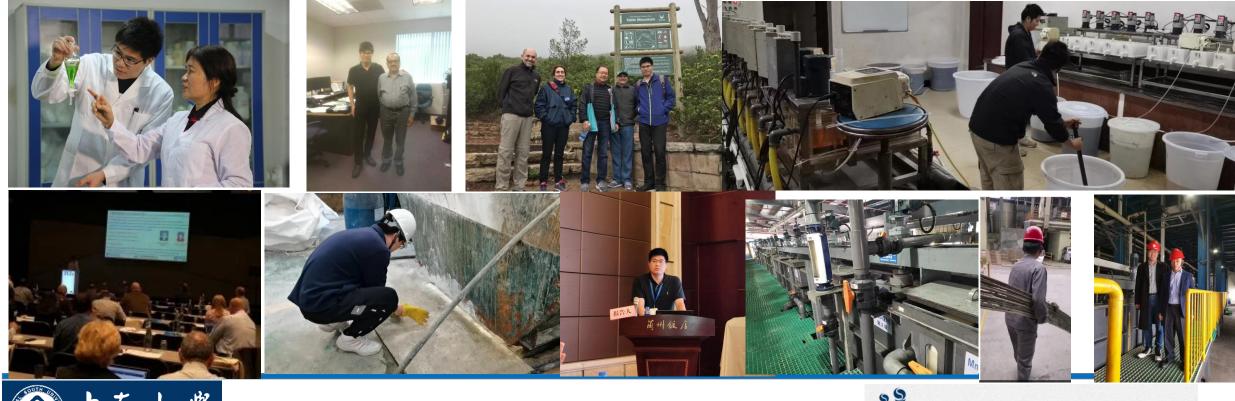
South University





Self introduction of Shengxi Wu/ Dr. Wu or Shengxi

Shengxi Wu, born in February 1991, from Fujian China, Ph. D. in engineering, now as a associate professor in Central South University (personal page: <u>https://faculty.csu.edu.cn/wushengxi/zh_CN/index.htm</u>) In the past about four years, published <u>41 papers, 20 patents and built up 12 leaching and extraction production lines</u> and finished 10 pilot scale line tests (treating >500L solution per day)







Selective co-extraction of Ni&Co from high Ca/Mg solutions



1. Research Background

2. Technological solution for Ni&Co co-extraction

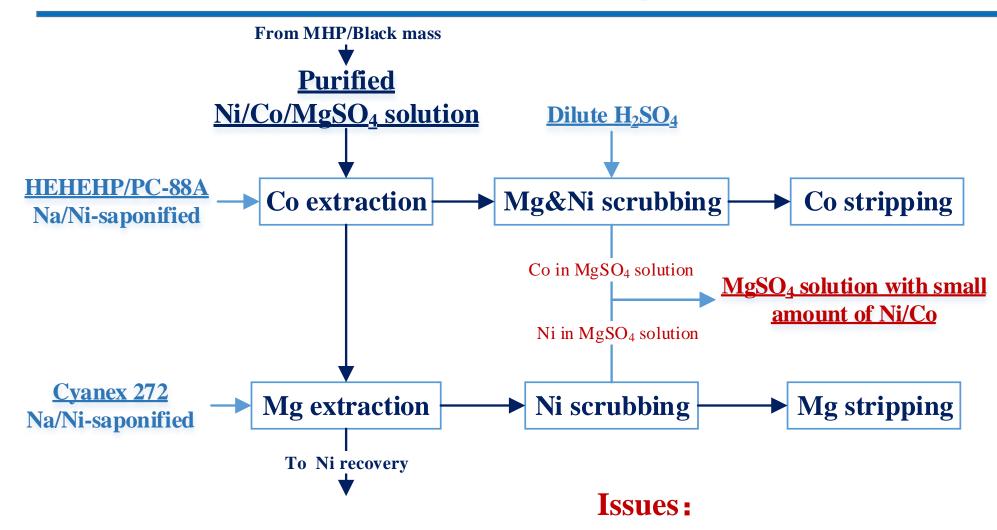
3. Experimental and industrial progress

4. Summary of Ni&Co co-extraction technology





1. Research Background



> Difficult in Ni/Co separation from massive Mg; High cost of sulfide precipitation, [Ni] unable to 1mg/L,

Toxic gas emission; Extra pressure oxidation leaching is needed;





plus: Basic conceptions and principles in solvent extraction

Reagent Cost Structure for metal recovery from spent LIBs

Seponification: $2NaOH+2\overline{(HL)_2} \leftrightarrow \overline{Na(L\cdot HL)_2} + H_2O$

Alkali for saponification;

Extraction: $Me^{2+} + 2 \overline{Na(L.HL)_2} \leftrightarrow \overline{Me(L\cdot HL)_2} + 2Na^+$

Acid for stripping and scrubbing

Scrubbing: $\overline{Me(L \cdot HL)_2} + 2H^+ \leftrightarrow Me^{2+} + 2\overline{(HL)_2}$

CaO/Na₂S for trace heavy metal and As/P/F precipitation in wastewater

Stripping: $\overline{Me(L \cdot HL)_2} + 2H^+ \leftrightarrow Me^{2+} + 2\overline{(HL)_2}$

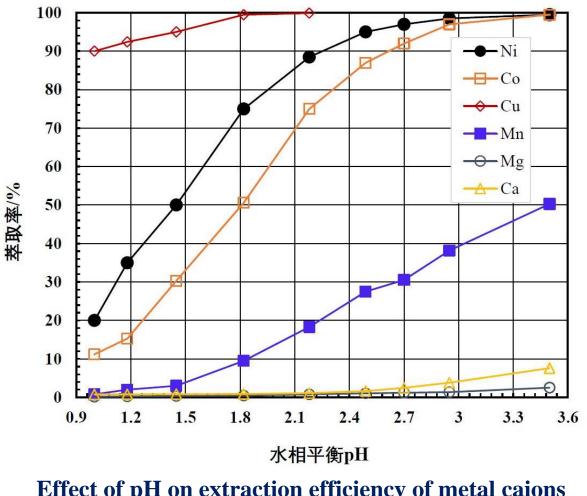
> NaOH for pH adjusting (neutralization)

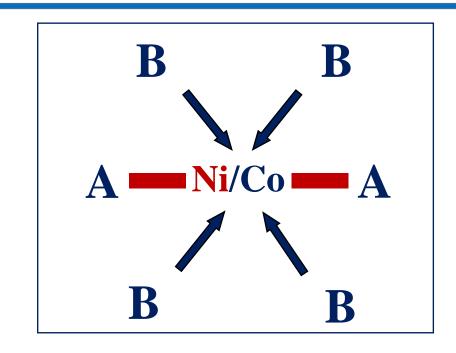
Principle 1st: Extracting small amount of elements from huge body element is economic





1. Extraction basis for Ni/Co from high Ca/Mg concentration solutions





A: Cation exchanger; **B:** Steric hindrance provider

Effect of pH on extraction efficiency of metal caions 20% HBL116 ; Feed (g/L) : 1.71 Ni²⁺, 0.12 Co²⁺, 15.7 Mg²⁺,

0.11 Ca²⁺, A/O 1:1; 30 °C



Ni and Co can be selectively extracted by HBL-116 (HT-059) from high Ca/Mg concentration solutions



1.1 batch extraction experiments

Five-stage countcurrent extraction of Ni/Co from high Ca/Mg concentration solutions

Elements	Со	Ni	Mg	Ca
Feed (g/L)	0.115	1.68	15.7	0.112
Raffinate (g/L)	0.0009	0.0038	16.1	0.127
Loaded organic (g/L)	0.141	4.36	0.0364	0.0041
Extraction efficiency (%)	99.8	99.8	0.093	1.47

Stripping efficiency of Ni/Co from loaded organic by 0.75mol H₂SO₄ via 5 stages

Elements	Со	Ni	Mg	Ca
Scrubbed organic (g/L)	0.140	4.35	0.0067	0.0011
Stripping solutions (g/L)	1.41	42.2	0.0801	0.0203
Regenerated organic (g/L)	0.0001	0.0131		
Stripping efficiency (%)	99.5	99.9		
Impurity removal efficiency (%)			99.97	99.26



Organic : 20% HBL116 + 80% diluent; Feed (g/L) : 0.115 Co²⁺, 1.68 Ni²⁺, 15.7 Mg²⁺, 0.112 Ca²⁺, O/A =1:2.5;



30 °C; saponification ratio: 50%; stripping O/A 10:1

1. Extraction basis for Ni/Co from high Ca/Mg concentration solutions

Five-stage countcurrent extraction of Ni/Co from high Ca/Mg concentration solutions

Rows	Co in raffinate (g/L)	Ni in raffinate (g/L)	Mg in raffinate (g/L)	Equibrium pH
1	1.723	0.903	13.17	3.48
2	0.007	0.003	13.10	4.43
3	0.005	0.001	13.12	4.33
4	0.0008	0.0005	13.15	4.32
5	ND	ND	13.11	4.36
6	ND	ND	13.15	4.29
7	0.0009	0.0004	13.18	4.31
8	0.0012	0.0006	13.35	4.28
9	0.0014	0.0008	13.40	4.23
10	ND	0.0005	13.19	4.26
11	0.0011	0.001	13.12	4.31
12	0.001	0.0004	13.15	4.35
13	0.0009	0.0008	13.11	4.32

E_{Ni}>99.99%, E_{Co}>99.99%



Feed: Co 8.50 g/L; Ni 9.70 g/L; Mg 14.2 g/L; pH 5.0

20% (v/v) HBL116, O/A=2.8:1, 225 rpm, t=10 min, T=25°C



Phase segregation performance



Photos of phase segregation row 13 (50s after stop mixing), stage 1st to stage 5th, excellent phase segregation excellent

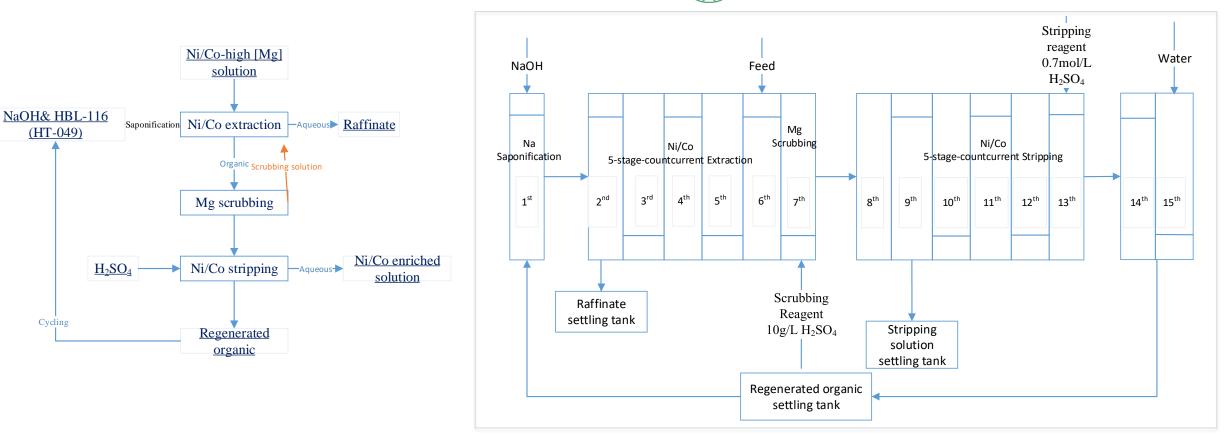




2.1 Guangdong Fangyuan New Material Group Co., LTD



着東芳源新材料集團服份有限公司 Guangdong Fangyuan New Materials Group Co., Ltd.



Production Lines: Selectively extraction of Ni/Co from ${\rm MgSO_4}$ solutions by HBL-116(HT-059). (launched in June 2020)







Pilot test & Production Lines: Selectively extraction of Ni/Co from MgSO₄ solutions by HBL-116(HT-059). Feed solution treating Scale: 400m³/day (launched in June 2020)

2.2 Continuous Running Data

		萃取		萃	洗	反	萃	反	洗	皂	化		萃余	液				反	萃液		
时间	有机	水相	萃余	流量	旺口	流量	反萃	流量	出口	皂化	碱量	1	Ni	C	0	I	Ni		Co		Mg
#11-3	L/h	L/h	рН	初重 L/h	рН	初重 L/h	рн	初重 L/h	рН	率/%	响鱼 L/h	mg/L	萃取率 /%	mg/L	萃取率 /%	g/L	反萃率 /%	g/L	反萃率 /%	g/L	除去率
01-09:00	3600	2800	6.76	150	5.15	750	1.1	530	1.55	70	63	未测出	99.99	0.8	99.94	3.81	99.95	11.55	99.99	0.023	>99.99%
01-11:00	3600	2800	6.64	150	5.2	800	0.93	560	1.77	70	63	未测出	99.99	0.1	99.99	3.98	99.97	10.33	99.99	0.014	>99.99%
01-13:00	3300	3300	6.98	150	5.47	800	0.93	560	1.82	73	60	0.4	99.99	0.3	99.97	3.12	99.98	10.15	99.99	0.007	>99.99%
02-09:10	3500	3500	7.55	150	5.68	800	0.99	560	1.92	71	62	0.07	99.97	0.8	99.94	3.19	99.97	10.76	99.95	0.010	>99.99%
02-11:00	3500	3500	6.98	150	5.61	800	1.01	560	1.91	70	65	未测出	>99.99	未测出	99.99	2.8	99.52	9.26	99.96	0.008	>99.99%
02-13:00	3700	3500	6.74	150	5.66	800	1.01	560	1.81	70	65	未测出	>99.99	未测出	99.99	4.14	99.52	11.88	99.96	0.012	>99.99%
03-09:00	3900	3500	6.29	150	5.22	800	0.96	560	2.13	70	68	0.06	99.98	0.27	99.98	4.29	99.89	14.5	99.97	0.012	>99.99%
03-11:00	3900	3500	6.27	150	5.25	800	1.02	560	1.97	70	68	0.74	99.72	0.19	99.99	3.84	99.87	15.74	99.97	0.013	>99.99%
03-13:00	3900	3500	6.57	150	5.49	800	1.17	560	2.09	70	68	0.42	99.85	0.13	99.99	3.24	99.86	15.19	99.97	0.006	>99.99%
04-09:00	3900	3500	6.31	150	4.85	750	1	530	2.02	70	68	0.43	99.84	0.35	99.97	4.29	99.86	14.5	99.97	0.012	>99.99%
04-11:00	4200	3500	6.27	150	4.9	750	1	530	1.82	70	73	0.74	99.03	0.19	99.88	3.84	<mark>99.8</mark> 7	15.74	99.97	0.013	>99.99%

表 3 HB116 镁水深净线皂化-萃取-萃洗-反萃-反洗连续稳定运行试验结果



The content of Ni/Co in raffinate was kept less than 1mg/L after 4 days of test running



2.2 Continuous Running Data

		萃取		萃	洗	反	萃	反	洗	皂	化		萃余	液				反	萃液		
时间	有机	水相	萃余	流量	出口	流量	反萃	流量	出口	皂化	碱量	1	Ni	С	0	N	li		Co		Mg
	L/h	L/h	рН	L/h	рН	L/h	рН	L/h	рН	率/%	L/h	mg/L	萃取率 /%	mg/L	萃取率 /%	g/L	反萃率 /%	g/L	反萃率 /%	g/L	除去率
25-09:00	<mark>6</mark> 500	2000	5.05	100	3.68	900	1.24	630	1.89	70	113	未测出	100	0.48	99.89	25.1	99.95	7.56	99.88	0.008	>99.99%
25-13:00	6200	2200	5.31	100	3.84	950	1.12	660	1.76	70	105	0.68	99.85	0.58	99.87	23.77	99.96	7.77	99.89	0.005	>99.99%
26-17:00	5600	2200	5.53	100	3.77	920	0.73	660	1.75	70	98	未测出	100	未测出	99.99	17.8	99.97	5.34	99.92	0.015	>99.99%
26-19:00	5300	2200	5.46	120	3.98	900	0.71	630	1.82	70	92	未测出	100	未测出	99.99	17.78	99.97	6.23	99.92	0.015	>99.99%
26-21:00	4700	2200	5.21	130	3.97	870	0.61	609	1.79	70	81	0.79	99.88	0.26	99.97	15.37	99.97	4.86	99.92	0.012	>99.99%
27-09:00	4700	2500	7.78	150	4.27	810	0.71	560	1.91	70	81	0.26	99.99	0.35	99.95	20.99	99.97	6.15	99.92	0.009	>99.99%
27-11:00	4700	2700	5.86	150	4.44	810	0.82	560	1.89	70	81	0.28	99.99	0.36	99.95	21.53	99.96	7.98	99.92	0.02	>99.99%
28-13:00	4700	3000	-	170	4.15	810	1.01	560	1.81	70	81	0.88	99.86	0.31	99.98	23.91	99.96	10.79	99.92	0.012	>99.99%
30-09:00	4800	2500	5.97	170	4.21	810	0.88	810	1.83	70	83	未测出	100	0.73	99.96	16.9	99.95	10.89	99.92	0.014	>99.99%
30-11:00	5100	2700	5.46	170	4.33	780	0.73	780	1.76	69	88	未测出	100	0.59	99.96	15.39	99.95	10.71	99.94	0.005	>99.99%
30-13:00	5100	2700	5.43	170	4.31	750	0.72	750	1.7	69	88	未测出	100	未测出	99.99	16.61	99.95	12.56	99.94	0.016	>99.99%
平均	4671	2935	6.09	177	4.34	774	0.98	556	1.80	69	81	0.40	100	0.49	99.96	12.89	99.92	14.36	99.94	0.081	>99.99%



The content of Ni/Co in raffinate was still kept less than 1mg/L after 45 days of production running



2.2 Continuous Running Data

Elements	Ni	Со	Mg
Feed (g/L)	0.1~3	1~5	25~60
Stripping (g/L)	1~4.5	10~15	1.07
Raffinate (mg/L)	<1	<1	>32
Extraction efficiency/%	99.9	99.9	
Mg removal efficiency/%			99.7

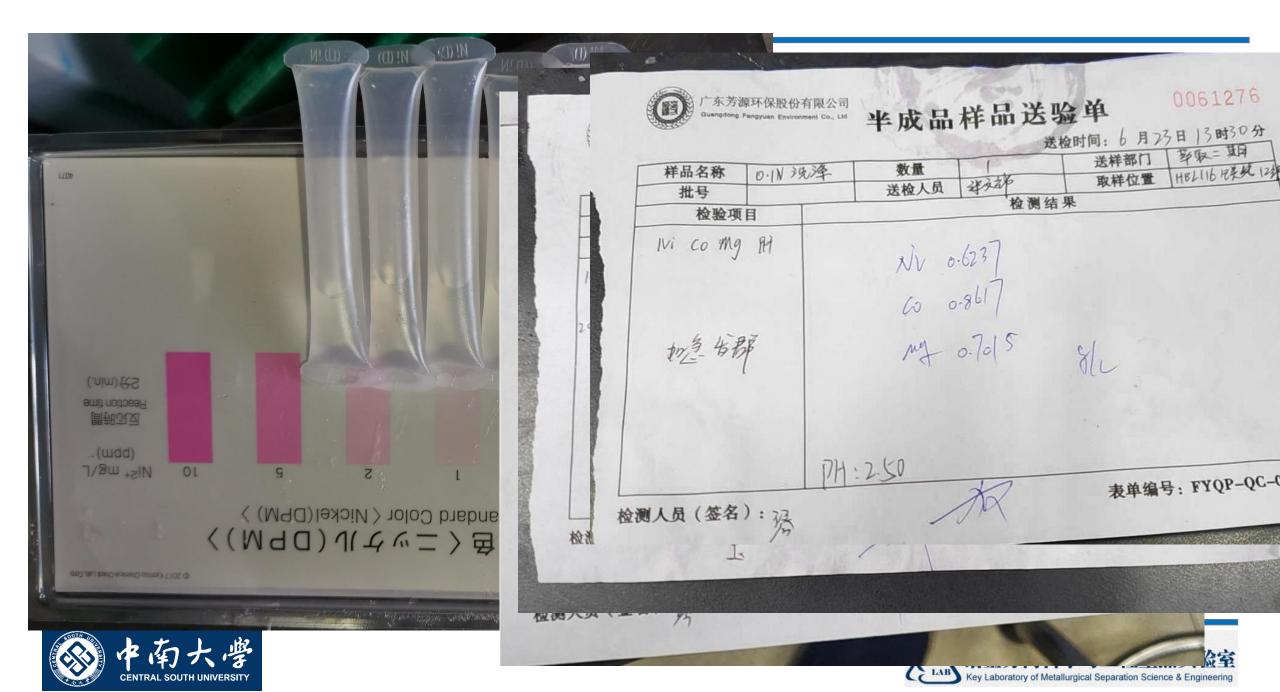


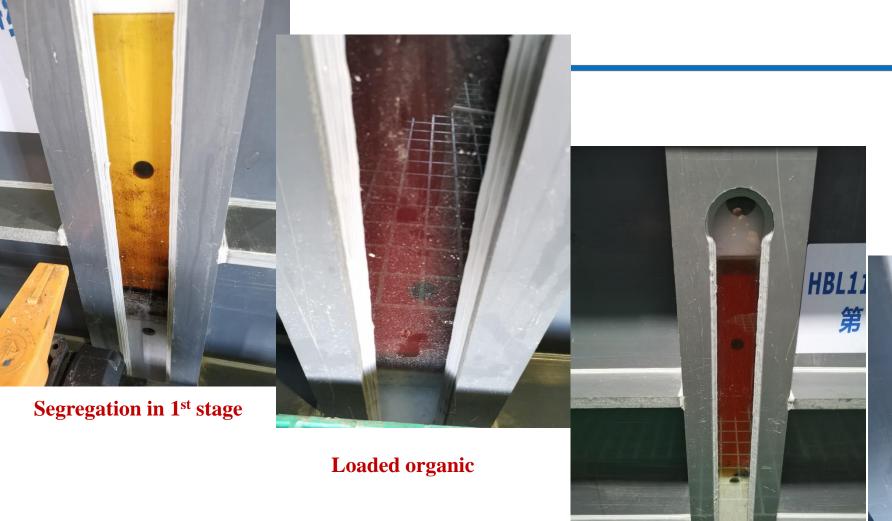
Oil content in raffinate: 8.14-16.06ppm, COD: 59.43-96.68ppm

Oil content in stripping: 4.19-10.97ppm, COD: 28.04-40.67ppm









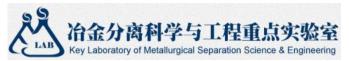
Regenerated organic

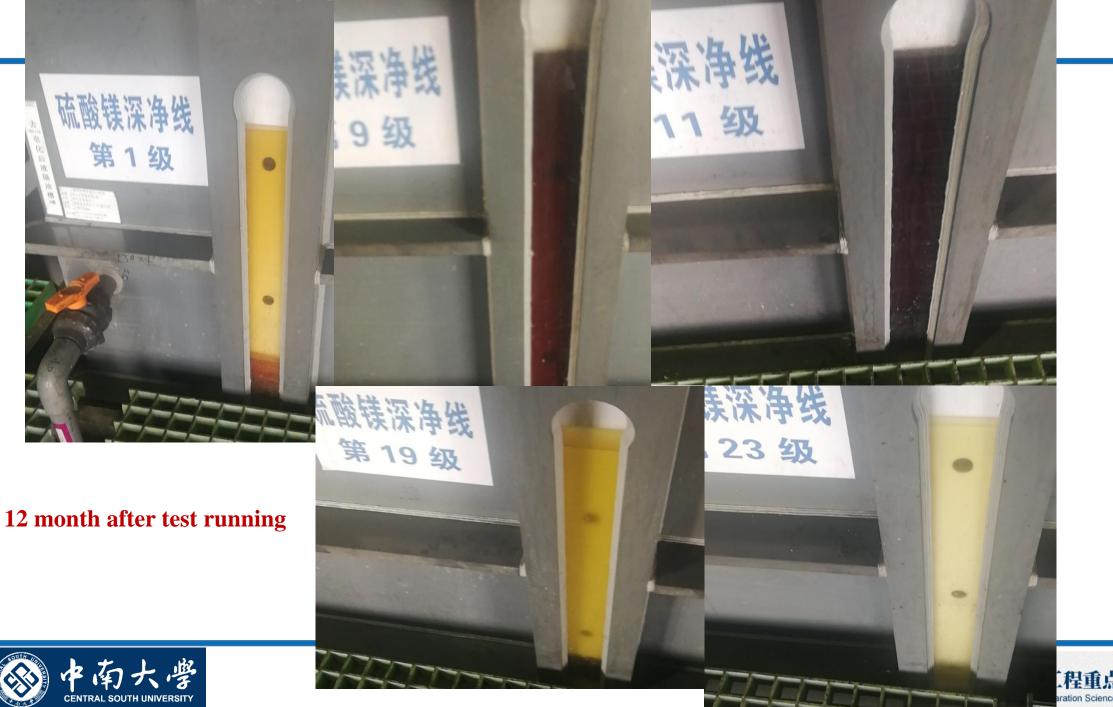
3 month after test running

Stripping segregation









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Reagent consumption for Ni/Co extraction from high Ca/Mg solutions

	Sulfuric ac	cid consumpti	on (98%H ₂ SO ₄)	Alkali (100%NaOH)			
Reagent	Scrubbing Acid	Stripping Acid	Acid in total	Saponified alkali	Total		
Theoretical consumption (t/t Ni&Co)		1.66		1.36			
Dosage (kg/m ³) feed solution	0.59	19.41	20.01	10.04	10.04		
Dosage (t/t) Ni/Co metal	0.09	2.92	3.01	1.66	1.66		



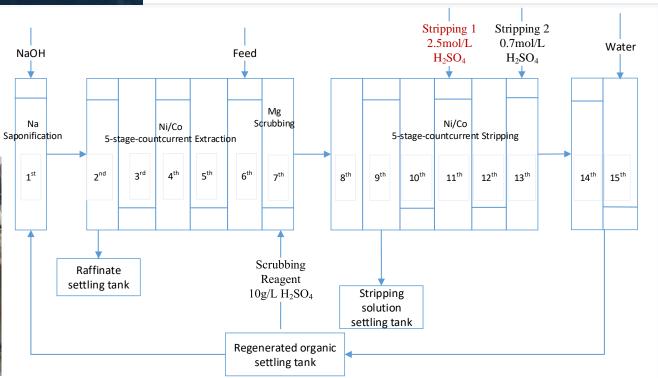


2.2 Huayou Cobalt Group Co., LTD ---pilot test



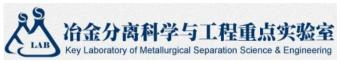
Goal: Ni+Co in raffinate <20ppm; Stripping solution Ni>80g/L, Mg<10ppm Ammonia<100ppm;





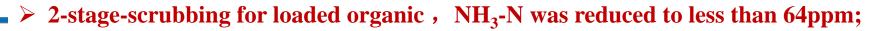
	Saponi	fication	extraction	Scrubbing	Stri	pping	scrubbin g			Raffi	nate			S	Stripping	solutio	on	
Flow	Organic	Alkali	Feed	0.3N acid	5N	1.5N	Water	Element	Ni	Co	pН	TOC	Ni	Co	Mg	pН	NH ₃ -N	TOC
rate(L/h)) 18	0.72	48	acia	acid	acid 0.96	3	g/L	0.009	0.02	4.25	0.025	87.52	0.75	0.0032	1.31	0.046	0.036





			Aqueous				Oragnic	
	Со	Mg	Ni	pH	NH ₃ -N	Со	Mg	Ni
萃取一	0.017	15.39	0.0062	4.26		0.31	4.66	0.49
萃取二	0.13	17.14	0.07	4		3.55	1.25	3.56
萃取三	1.73	16.08	1.74	3.33		2.65	0.14	6.35
萃取四	1.43	15.34	3.58	3.15		1.07	0.048	8.13
萃取五	0.87	14.91	4.66	3.18		0.44	0.046	9.83
洗涤一	1.51	0.89	10.51	2.82	0.36	0.42	0.0087	8.77
洗涤二	1.34	0.017	12.28	2.7	0.064	0.35	0.0026	9.00
洗涤三	0.99	0.0034	11	2.65	0.054	0.24	0.00086	9.11
洗涤四	0.58	0.0011	12.25	2.58	0.049	0.17	0.0022	9.04
洗涤五	0.27	0.0017	7.92	2.15	0.032	0.1	0.00084	7.70
反萃一	0.78	0.00098	83.23	1.34	0.055	0.047	0.00051	6.98
反萃二	0.32	0.00085	79	0.85	/	0.026	0.00041	5.46
反萃三	0.1	0.00062	51.05	0.2	/	0.029	0.00037	1.48
反萃四	0.07	0.00097	27.06	0.2	/	0.069	0.00043	1.4
反萃五	0.031	0.00073	12.97	0.1	/	0.047	0.00069	1.3

> Ni Raffinate is less than 50ppm (meet the low limit concentration requirement);





> Ni concentration in stripping solution is about 80g/L with Mg <10ppm





Reagent consumption for Ni/Co extraction from high Ca/Mg solutions

	Liquid alkali (32% NaOH)/kg	$5N H_2 SO_4 /m^3$	Feed/m ³	Solvent loss /kg	Diluent loss /kg
Reagent t/t Ni	5297	9.89	32.17	2.81	4.22





HBL-116 (HT-059) extraction production line for Ni selective extraction from Co extraction raffinate produced for the Cu-Co ore, will be launched June 2024, Brunp, Yichang, Hubei, Chine

 $(1462m^{3}/d)$



Ni(g/L)	Co(g/L)	Mg(g/L)	NH₄⁺(g/L)	SO₄²¯(g/L)	油分(mg/L)	Cod(g/L)	рН
1.12	0.10	3.38	27.48	93.04	40-60	1.2	5.0-5.5

	HBL 116 (25%) 萃镍线出口水相金属浓度											
物料 元素	Ni(g/L)	Co(g/L)	Mg(g/L)	油分(mg/L)	Cod(g/L)	рН						
萃余液	< 0.005	< 0.005	3.26	40-60	1.20	3.5-4.5						
反镍液	40.42	3.61	1.22	50-60	0.60	0.5-1.5						



HBL 116 (25%) 萃镍线级数									
萃取功能	铵皂	萃余液澄清	萃取	洗铵镁	有机澄清	反萃	洗酸	有机澄清	总级数
级数	1	2	5	3	1	6	3	1	22
萃取槽混合室规格	φ1.8×2.65m	/	φ1.8×2.65m					/	/
萃取槽澄清室规格	7*2.1*1.3	7*4.0*1.3	7*4.0*1.3 7*2.1*1.3				7*2.1*1.3	/	

Lanzhou Jintong Energy Storage Power New Materials Co., LTD, Austin Elements (USA) pilot test for

Ni&Co coextraction from Li/Ni/Co/Mg mixed solutions, Guizhou Daong Huicheng Co., LTD using HBL-

110 (HT-059-2nd) to selectively co-extract Ni/Co/Mn from Ca/Mg/Fe/Al/Ni/Co/Mn mixed solutions







3. Industrial application of Ni/Co extraction from Ca/Mg/Fe/Al mixed solutions



Direct selectively extraction of Ni from acid leach solutions of stainless steel sludge, Jiuli Co. Ltd. , Zhejiang Province NiSO₄: 800 ton/a, put into production in Feb, 2018





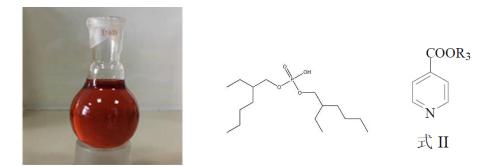
Electroplating sludge contains massive various impurities with a high concentration

Element	Zn	Ni	Ca	Mg	Al	Fe
Con.	~20g/L	5~6g/L	~0.5	~0.3g/L	~0.1g/L	<0.1g/L

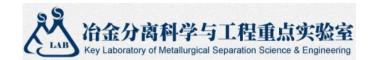


production line5: <u>Direct</u> extraction of Ni from the leaching solution of EPS in Ruijin Shengyuan Environmental Protection Technology Co., LTD.

Using HBL110 as the extractant for Ni



Components in stripping solutions: Ni 98g/L Zn, Ca, Mg<10ppm; Al, Fe<30ppm





3. Industrial application of Ni/Co extraction from Ca/Mg/Fe/Al mixed solutions





NiSO₄: 800 tonne/a Mixer: 0.42m³; Put into production on Aug. 2014

DSX of Ni from acid leach solutions of electroplating sludge using HBL110 (HT-059-2nd) Shuangneng, Ningbo, Zhejiang Province





3. Industrial application of Ni/Co extraction from Ca/Mg/Fe/Al mixed solutions

Guizhou DalongHuicheng Co. Ltd. Acid leaching solution of heavy metal sulfide

precipitate that contains high Ca/Mg/Mn and considerable Ni/Co

Feed solution							
Element Mn		Со	Ni	Ca	Mg		
Con. (g/L)	50~80	4~5	2~3	0.1~0.2	0.1		



Using HBL110 (HT-059-2nd) directly prepare pure NCM mixed solution

Stripping solution							
Element	Mn	Со	Ni	Ca	Mg		
Con. (g/L)	~17	~25	~10	<5ppm	<5ppm		

The first and the only plant that recover NCM without separation in China





Summary for direct solvent of Ni/Co from high Ca/Mg solutions:

- > Short process omit of sulfide precipitation and pressure oxidation leaching operations
- **Excellent Ni/Co recovery >99%,** illustrating a **>10%** increase in Ni/Co recovery;
- **Significantly reduce scrubbing solution** (20~60g/L Mg, 0.5~10g/L Ni/Co);
- Low Cost recovery method for Ni/Co from high Ca/Mg (even with Fe/Al solutions)
 HBL116 (HT-059) for Ni/Co separation from Ca/Mg, HBL-110 (HT-059-2nd) from Ca/Mg/Fe/Al

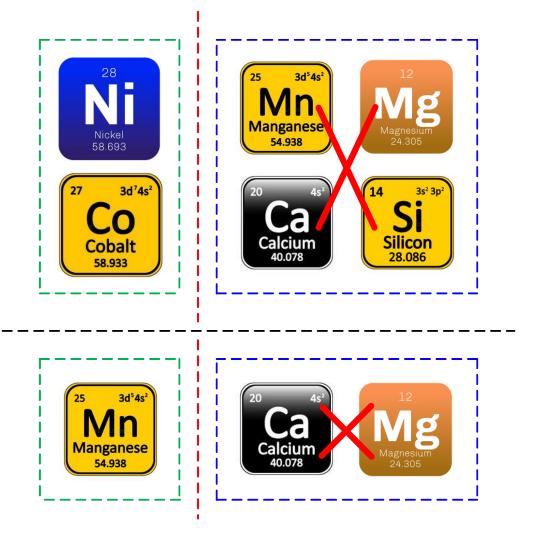
Environmental friendly, Ni/Co in raffinate was reduced to <1mg/L



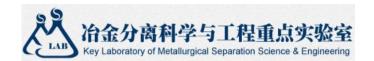


4. Summary and Thanks











Lab of Metallurgical Separation Science and Engineering

Central South University





Qinggang Li Prof.

Zuoying Cao Prof. Mingyu Wang Prof.

Li Zeng Prof.

Dr. Zuoying Cao

Dr. Shengxi Wu





Chongjie Gao Prof.

Cheng Chuyong Prof. (LAB) 印版刀內科学与工程重点实验室 Key Laboratory of Metallurgical Separation Science & Engineering

4, Summary

□ Core conception: "separation as the key method, application is the final goal,
 achieve resource sustainable development via green and high efficient utilization"



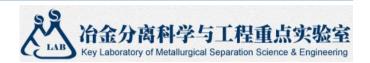


Technology development



Engineering transformation and application





4. Summary- International project solution provider in hydrometallurgy



Hydrometal Tech: International website for our group project

Pioneering Sustainable Solutions provider in Battery Recycling

Hydrometal Tech was founded from a vision to revolutionize the battery recycling technologies in industry. With over 50 years of combined experience in hydrometallurgy, our founders, a group of esteemed professors and engineers, saw the potential to make a substantial impact on the environment and the economy. Our journey began with a commitment to excellence, innovation, and sustainability, which continues to be the bedrock of our operations today.

Website: https://hydrometaltech.com/

E-mail: shengxiwu@hydrometaltech.com





4. Summary and Thanks

Thanks for your time and question are welcomed





